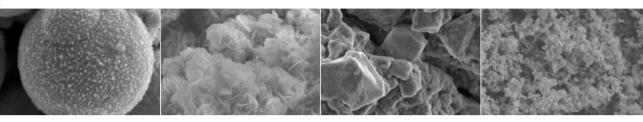
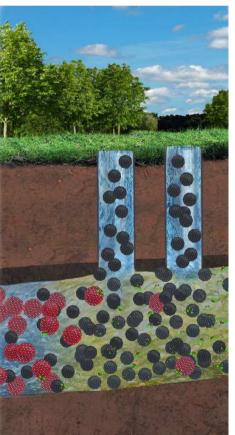
# ADVANCED WATER TREATMENT TECHNOLOGIES







NANOMATERIALS
IN ENVIRONMENTAL APPLICATIONS











# **GROUNDWATER REMEDIATION**

REDUCTIVE TECHNOLOGIES BASED ON NANOSCALE ZERO-VALENT IRON



#### HIGHLY EFFICIENT IN REMOVAL OF

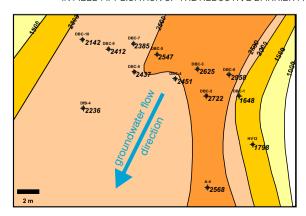
- Inorganic compounds like Arsenic, Barium, Chromium, Selenium, Uranium, Copper, Lead, Cadmium, Mercury, Nickel (and many other heavy metals), sulphates, nitrates, perchlorates, etc.
- Organic compounds like polychlorinated methanes, ethanes, ethenes, benzenes, etc.

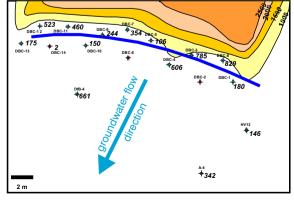
#### PROPERTIES OF FE<sup>0</sup> NANOPARTICLES

- > High redox potential (electron donor)
- > High surface area (20-25 m<sup>2</sup>/g)
- > Narrow particle size distribution (average of 50 nm) allowing the particles to migrate in groundwater conditions
- > Pyrophoric or air-stable character (powder or aqueous dispersion)
- > Magnetically separable
- > High-reactivity and long-term performance
- > Possibility to combine various properties in composite materials
- Ideal for remediation of large areas or for construction of permeable reactive barriers
- > Environmentally friendly material
- > Low cost comparing to conventional technologies

#### APPLICATION EXAMPLE

IN FIELD APPLICATION OF THE REDUCTIVE BARRIER AT MARS SVRATKA (CZECH REPUBLIC) IN 2012-2014)







# SURFACE AND DRINKING WATER TREATMENT

OXIDATIVE TECHNOLOGIES BASED ON IRON IN HIGH VALENCE STATE (FERRATES IV,V,VI)



#### HIGHLY EFFICIENT IN REMOVAL OF

- Inorganic compounds such as Arsenic, Uranium, Zinc, Copper, Nickel (and many other heavy metals), metal cyanides, phosphates, sulfates, etc.
- Organic compounds such as hormones, pesticides, pharmaceutical compounds, etc.
- > Biological pollutants such as Cyanobacteria

#### PROPERTIES OF FERRATE MATERIALS

- > High oxidative potential (electron acceptor) due to Fe in high valence state
- > Highly effective against wide spectrum of pollutants
- Disinfecting effect against various microorganisms (Cryptoporidium, Coliforms, Salmonella, etc.)
- Coagulation and flocculation capacity
- > Increasing pH value
- > High reactivity and low reagent consumption
- Reaction products are benign iron oxides with high surface area which is advantageously useful for additional sorption
- > Environmentally friendly material
- > Low cost comparing to conventional technologies



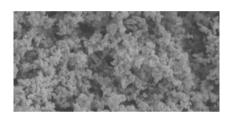
# **AVAILABLE PRODUCTS**





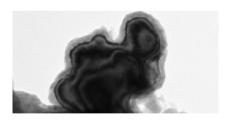
#### NANOFER 25P

- > Fe<sup>o</sup> nanoparticles (powder) without surface modification
- > Highly reactive and pyrophoric Fe<sup>o</sup> nanoparticles



#### NANOFER 25S

- Aqueous dispersion of Fe<sup>o</sup> nanoparticles modified by biodegradable organic stabilizer
- > Highly reactive with large scale of pollutants
- > Excellent migration properties

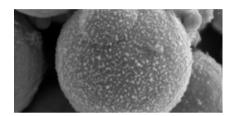


#### NANOFER STAR

- > Surface-stabilized Fe<sup>o</sup> nanoparticles (powder)
- Safe storage, transport, handling and applications compared to non-stabilized Fe<sup>o</sup> nanoparticles
- When activated, the Fe° nanoparticles are highly reactive with reducible pollutants
- Possibility to combine with biodegradable organic stabilizer for better migration performance

#### FE<sup>0</sup>-BASED COMPOSITES

- Micro-Fe<sup>o</sup> & nano-Fe<sup>o</sup> composite material combining long-term reactivity of microparticles with rapid effect of nanoparticles
- Nano-Feº & carbon composite material combining reductive/magnetic properties of Feº nanoparticles with sorption properties of carbon (active carbon, carbon black, nanotubes, biochar etc.)



#### **FNVIFFR**

- Product based on Ferrate (IV, V or VI, or their combination) with outstanding oxidation capacity
- Oxidation/disinfection properties are accompanied with high sorption capacity of reaction products
- > Soluble material stable in solutions at high pH
- > Stable in solid form at dry conditions

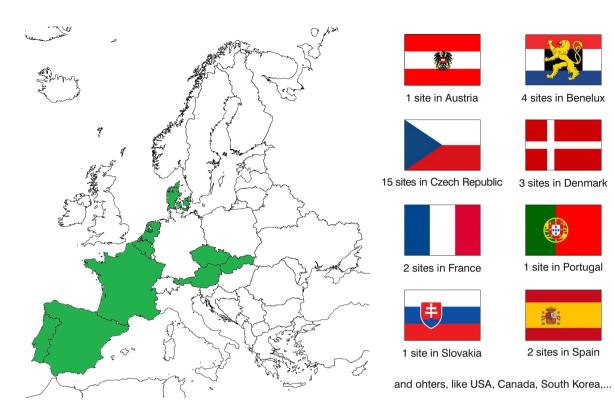




#### **AND MUCH MORE...**

- We are able to offer custom modified Fe-based nanomaterials
- > Any additional information on request

### APPLICATION REFERENCES



## SCIENTIFIC REFERENCES

- ACCOUNTS
- Science & lectinology

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- Soukupova, J. et al. Highly concentrated, reactive and stable dispersion of zero-valent iron nanoparticles: Direct surface modification and site application. Chemical Engineering Journal 262, 813-822, 2015
- Sharma, V. K. et al. Ferrates: Greener Oxidants with Multimodal Action in Water Treatment Technologies. Accounts of Chemical Research 48, 182-191, 2015
- > Jarosova, B. et al. Can zero-valent iron nanoparticles remove waterborne estrogens? Journal of Environmental Management 150, 387-392, 2015
- > Prucek, R. et al. Ferrate(VI)-Prompted Removal of Metals in Aqueous Media: Mechanistic Delineation of Enhanced Efficiency via Metal Entrenchment in Magnetic Oxides. Environmental Science & Technology, 49, 2319-2327, 2015
- > Filip, J. et al. Anaerobic Reaction of Nanoscale Zerovalent Iron with Water: Mechanism and Kinetics. Journal of Physical Chemistry C 118, 13817-13825, 2014
- Prucek, R. et al. Ferrate(VI)-Induced Arsenite and Arsenate Removal by In Situ Structural Incorporation into Magnetic Iron(III) Oxide Nanoparticles. Environmental Science & Technology 47, 3283-3292, 2013
- > Zboril, R. et al. Treatment of chemical warfare agents by zero-valent iron nanoparticles and ferrate(VI)/(III) composite. Journal of Hazardous Materials 211, 126-130, 2012
- Marsalek, B. et al. Multimodal Action and Selective Toxicity of Zerovalent Iron Nanoparticles against Cyanobacteria. Environmental Science & Technology 46, 2316-2323, 2012
- > Klimkova, S. et al. Zero-valent iron nanoparticles in treatment of acid mine water from in situ uranium leaching. Chemosphere 82, 1178-1184, 2011

INDUSTRIAL PARTNERS:





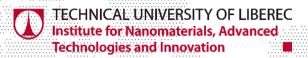








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